Electrical properties of EVA/LDPE blends irradiated by high energy electron beam

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Abstract. A series of low-density polyethylene (LDPE) blends with different amounts of poly(ethylene-vinyl acetate) (EVA) was prepared and irradiated with 10 MeV electron beam in the range of 50–380 kGy at room temperature in air. Some electrical properties of the blends such as surface and volume resistance, breakdown voltage, permittivity and loss factor were studied as a function of the EVA content and the radiation dose. It was revealed that the surface resistance and volume resistivity of the blends reach a maximum at a 170 kGy dose of radiation and 30 wt% of EVA. There is no considerable change at breakdown voltage permittivity and loss factor of the irradiated samples; however, the permittivity and the loss factor of the blends increase significantly with increasing amount of EVA contents.

Key words: EVA/LDPE blends • high energy electron beam • electrical properties • surface resistances • permittivity (dielectric constant) • loss factor

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Introduction

Polymers have found extensive applications in cables and capacitors as dielectric materials [24, 27, 29, 39]. Mixing of poly(ethylene-co-vinyl acetate) EVA and semi-crystalline polymers such as low-density polyethylene (LDPE) gives composite materials having better flexibility, toughness and high resistance to environmental stress cracking due to the increase of the adhesive strength at the matrix-rubber particle interface [1, 14, 28, 30].

The treatment of such composition with high energy radiation leads to various changes in their structure and properties [3, 7]. Electron beam (EB) processing has been demonstrated, to be a very effective means of improving and use properties of various polymers [9, 23]. Until now, many investigations have been reported on the effects of low, medium and high energy radiation on mechanical, electrical and thermal properties of EVA/LDPE blends [5, 16, 20, 22]. Nevertheless, studies of the effect of high energy electron beam on electrical properties of the blended EVA/LDPE samples have received less attention.

In a previous work [31], we reported the effect of high energy electron beam on mechanical and thermal properties of EVA/LDPE blends. In this work, the effect of 10 MeV electron beam on some electrical properties such as: surface resistance, volume resistivity, permittivity (dielectric constant), loss factor and voltage breakdown of these blends is reported.

Experimental

LDPE (LH-0075) with a density of 0.921 g·cm⁻³ was obtained from Bandar Emam Petrochemical Company (Iran). EVA containing 18% vinyl acetate was purchased from BP Co. UK. The EVA/LDPE blends in ratios of 0.0, 10, 20, 30 and 100 wt% of EVA were melt-mixed and then molded to flat sheets (2 ± 0.05 mm thickness) using hot pressing at a temperature of 180°C for 5.0 min under 20 MPa pressure and the sample were then cooled to room temperature. Radiation of the sample sheets was carried out at room temperature in air by a high energy electron accelerator (TT200, Yazd/Iran Radiation Processing Center) under various radiation doses ranging from 50 to 380 kGy.

Surface and volume resistance were measured at room temperature by a teraohmmeter, made by CEAST Company.

A dielectric loss measurement system, model TRS-10T (ANDO Electric Co., Japan) was used for determining the dielectric constant and the dielectric loss tangent of the samples. Frequency was tuned at 1 MHz in the experiments.

The voltage breakdown of the samples was measured by a Dielectric Rigidity system P/N 6135.053, CEAST Company, made in Italy.

Results and discussion

Figure 1 compares the surface resistance of EVA/LDPE blended samples containing 10, 20 and 30 wt% of EVA with unblended LDPE and EVA as a function of irradiation dose. It is clear from Fig. 1 that the surface resistance of the unblended EVA samples is substantially lower than that of LDPE and also EVA/LDPE blends. In addition, it can be seen that the more content of EVA is used in the blend, the lower surface resistance is achieved. All samples (except unblended LDPE) showed an increase of surface resistance when irradiated by high energy electron beam up to 170 kGy of irradiation dose. Around the 170 kGy radiation dose, which is an optimum dose for the best cross-linking of the EVA/LDPE blends [31]; there is a noticeable increasing of surface resistance for all blends. For the radiation doses higher than 170 kGy, the surface resistance of the samples decreased



Fig. 1. Surface resistance vs. absorbed dose of 10 MeV electron beam for (\Diamond) unblended EVA, (*) LDPE and EVA/LDPE blends at (\Box) 10, (\triangle) 20 and (×) 30 wt% of EVA.

abruptly (with the exception of LDPE), and then converged to each other.

With increasing EVA content in the composition up to 30 wt%, the maximum of the trace observed at 170 kGy is increased more rapidly than the blends with the smaller content of EVA component. In addition, the surface resistance of the unblended EVA sample is substantially smaller than that of the blended samples at high irradiation doses. With increasing radiation dose, however, the surface resistance of the LDPE remained unchanged. Similar trend is revealed in Fig. 2 in which alike the surface resistance curves, the volume resistivity of the blend containing 30 wt\% of EVA is higher than the other samples around the 170 kGy radiation dose. Nevertheless, the volume resistivity at higher doses for all cases are identical. The noticeable peaks in the surface and volume resistivity curves around 170 kGy may be related to higher crosslinking density at this dose of radiation which causes to increase the average molecular weight of polymer. Presence of much more cross-linking points can be considered as barrels to prevent the charge movement between polymer chains and thus increase the electrical resistance of cross-linked sample [4, 25]. This argument is in agreement with the Kolesov study on the PE in which he described the influence of molecular weight on morphological and electrical properties of PE, he also has shown that the increase of the molecular weight leads to an increase in the volume resistivity [17]. Also Marsacq et al. [21] have shown that the increase in molecular weight leads to an increase in charging ability of PE, i.e. the growth of a number of traps in material which prevent the charge movement inside the polymer and hence increase the electrical resistance. Therefore, it can be said that the accordance of electrical resistance peak of the blend with the 170 kGy irradiation dose is related to the increase of traps and molecular weight of polymer due to cross link formation.

At higher doses, the cross-linking process occurs along with chain scission reaction, probably this later is predominating [8, 35]. As a result, non homogeneities owing to molecular chain segmentation which are introduced into the polymer matrix and polar and ionic products due to irradiation can probably cause the resistance of samples to decrease [18, 26].

As can be observed from Figs. 1 and 2, the surface resistance and volume resistivity of the blends around



Fig. 2. Volume resistivity vs. absorbed dose of 10 MeV electron beam for (\Diamond) unblended EVA, (*) LDPE and EVA/ LDPE blends at (\Box) 10, (\triangle) 20 and (×) 30 wt% of EVA.



Fig. 3. Breakdown voltage vs. absorbed dose of 10 MeV electron beam for (\diamond) unblended EVA, (*) LDPE and EVA/ LDPE blends at (\Box) 10, (\triangle) 20 and (×) 30 wt% of EVA.

the 170 kGy increase with the content of EVA. Several studies have indicated that the gel content and the crosslinking density of EVA/LDPE blends at an optimum radiation dose increase with increasing EVA content [10, 31, 32], and the highest gel content is observed when the amount of EVA in the blended samples reaches 30 weight percent [10, 11]. So, we can state that the highest surface resistance and volume resistivity of the blends at 170 kGy and 30 wt% of EVA may be related to the highest cross-linking density.

In Fig. 3, the breakdown voltage of the samples against the radiation dose has been plotted. EVA samples undergo dielectric breakdown at voltages lower than LDPE and EVA/LDPE blends. In all cases no considerable variation of breakdown voltage is revealed when the radiation dose is increased. This may be attributed to the fact that increasing of the radiation dose leads to an increase in cross-linking degree in amorphous area which may be barrels to prevent the electrical breakdown path inside the polymer matrix [19, 25, 40]. On the other hand, increasing of the radiation dose increases the chain scission of the PE and leads to increase the carbonyl groups and also shallow trap centers [6, 15], so, this causes to increase the mobility of the charge carrier and finally decrease the breakdown voltage. The mutual effect of radiation, i.e. the increase of breakdown voltage due to cross-linking on the one hand, and the decrease of breakdown voltage due to chain scission, on the other hand, causes the breakdown voltage of the polymer sample to remain without considerable variation. Our assumption is that the above argument shows an opposite behavior, so it may be concluded that the radiation dose does not make effect so much on the breakdown variation. From Fig. 3, it can also be seen that the blends and EVA break electrically in voltages lower than LDPE samples which may be attributed to the presence of polar groups on the EVA back bone [34, 38].

Figures 4 and 5 show the dependence of the permittivity (dielectric constant) and the loss factor of the samples on the radiation dose, respectively. Figure 4 shows that the permittivity of the blends is smaller than that of EVA, and with increasing radiation dose the permittivity increases very slowly. One may conclude that the electron beam radiation disintegrates molecules and increases the number of dipoles, so that the higher radiation dose produces a larger permittivity [13, 33].



Fig. 4. Variation of the dielectric constant of (\diamondsuit) unblended EVA, (*) LDPE and EVA/LDPE blends at (\Box) 10, (\bigtriangleup) 20 and (×) 30 wt% levels of the former, with absorbed dose of 10 MeV electron radiation.

The permittivity of the polymer is representative of the various polarization phenomena that come into play when the polymer is subjected to an electric field. The overall polarization of a polymer, like the PE, is the sum of four terms: electronic, atomic, orientation and spacecharge polarization, among them the first two are intrinsic in nature and for non-polar polymer are important [2]. For polar polymers, both the atomic and electronic polarization are often negligible compared to orientation and space charge polarization [2]. It has been shown that there is a relationship between injecting depth and energy of electron radiation. For example, as the radiation energy of electrons is 15 MeV or 30 keV, the injecting depth can be approximately deduced as 4 mm and 50 µm, respectively [36, 37]. This means that 10 MeV electron beam can penetrate totally inside the 2 mm polymer sheet, so there will be very little space charges left in the bulk of the sample. So, it is the orientation polarization that has a major contribution to the permittivity of a polar polymer like the EVA. This arises owing to the presence of permanent dipoles on the back bone of the macromolecular chain. Since, LDPE is a non-polar polymer with smaller permittivity than EVA; its blends would show permittivity which is held between the permittivity of the LDPE and EVA depending on the content of EVA component in the blends. In other words, the more content of EVA in the blends, the larger amount of permittivity will be achieved. It is desirable to keep the capacitance of the



Fig. 5. Variation of the dielectric loss factor of (\diamondsuit) unblended EVA, (*) LDPE and EVA/LDPE blends at (\Box) 10, (\bigtriangleup) 20 and (×) 30 wt% levels of the former, with absorbed dose of 10 MeV electron radiation.

insulating material as minimal as possible when they are used as cables in industries. Furthermore, from Fig. 4 it can be seen that although the permittivity of the blends is smaller than that of EVA and larger than that of LDPE, however, the content of EVA in the blends (up to 30 wt%) and the radiation dose (up to 380 kGy) have negligible influence on the permittivity of the LDPE samples.

Figure 5 reveals that the loss factor of the LDPE polymer is smaller than that of EVA/LDPE blends and EVA which can be related to the presence of polar groups and heterogeneity nature of the blend samples. For non-polar polymers, the loss factor may, however, increase sharply due to the presence of additives, impurity concentrations and physical heterogeneity [12]. However, the loss factor of all samples remains without significant change with increasing radiation dose. This means that mixing of a polar polymer at low amount level can affect the electrical properties of the non-polar polymer, whereas the radiation dose cannot create considerable effect on the electrical properties of such polymers.

Conclusions

Effects of adding the EVA polymer and of high energy electron radiation on the electrical properties of lowdensity polyethylene have been investigated. The surface and volume resistance and also breakdown voltage of the blends prepared by mixing of various amounts of EVA with LDPE decrease compared to the unblended LDPE, while the permittivity and loss factor increase. The larger increase of the amount of EVA in the blends, the greater change of the cited electrical properties occur in the polymer sample. On the other hand, the breakdown voltage, dielectric constant and loss factor of all the blends remain relatively constant and the surface and volume resistance of the samples were affected significantly by increasing the radiation dose. So, the resistance of all blends increases when the radiation dose increases up to 170 kGy. The increase of the resistance of the blend containing 30% of EVA is larger than that of the other blend samples. Increasing the radiation dose over 170 kGy considerably reduces the surface and volume resistance of all samples.

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